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Modern Chemistry Chapter 13 • Ions in Aqueous Solutions and Colligative Properties Section 1Compounds in Aqueous Solutions • dissociation is the separation of ions that occurs when an ionic compound dissolves – see sample problem A on page 436 – do practice problem #1 on page 436 Solubility • precipitation reactions occur when combinations of ions in a solution have an extremely low solubility and a precipitate forms – see table 1 on page 437 of the textbook for general ...

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CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative Properties SECTION 2 PROBLEMSWrite the answer on the line to the left.

13 Ions in Aqueous Solutions and Colligative Properties

Holt Modern Chemistry Review CHAPTER 12 & 13: SOLUTIONS The following pages contain the bulk (but not all) of the information for the chapter 12 & 13 test. Focus on this content, but make sure to review class notes, activities, handouts, questions, etc. If you study this document and NOTHING else, you should at least be able to PASS the test. ***** Test items will be recall, examples, and/or ...

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Chemistry Chapter 13 Test Review multiple choice (20) relationship of moles of ions to molecules recognize net ionic equations definition of precipitation reaction use solubility guidelines to identify precipitates definitions of dissociation & ionization the hydronium ion formula & its anions definitions of weak & strong electrolytes colligative properties how nonelectrolytes, electrolytes, and nonvolatile solutes affect colligative properties calculate molality using freezing point ...

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CHAPTER 14 REVIEW Ions in Aqueous Solutions and Colligative Properties SECTION 14-1 SHORT ANSWER Answer the following questions in the space provided. 1. Use the guidelines in Table 14-1 on page 427 of the text to predict the solubility of the following

14 Ions in Aqueous Solutions and Colligative Properties

Modern Chemistry Chapter 13 Ions in Aqueous Solutions and Colligative Properties Section 1- Compounds in Aqueous Solutions dissociation is the separation of ions that occurs when an ionic compound dissolves see sample problem A on page 436 do practice problem #1 on page 436 Solubility precipitation reactions occur when combinations of ions in a solution have an extremely low solubility and a ...

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